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REMOVAL OF THE HARMFUL CARBONIC COMPONENTS FROM EXHAUST GASES OF TRANSPORT VEHICLES

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Abstract- The exhaust gases of transport vehicles and volatile organic components (VOC) are known as basic atmosphere contaminants. For abatement of these contaminants, we carried out studies on removal of the harmful carbonic components from air by oxidation in the presence of ozone over catalyst. We proposed a new approach for the atmospheric air cleaning the use of ozone together with the catalyst.

Keywords: Ozone, Polluted Air, Toluene, Oxidation, Volatile Organic Components (VOC).

I. INTRODUCTION

The well-known methods of air treatment from the pollutants are oxidation and catalytic oxidation, filtration and biofiltration, condensation, absorption, adsorption and finally our method of oxidation with ozone over a catalyst. It is shown [1-5] that ozone application in combination with the catalyst allows to remove the harmful carbon-containing components from the air.

Therefore, in this study the effect of reactor temperature, presence of ozone and porous material (helps to increase the contact area between polluted component and ozone) on pollutant removal has been studied. High energy, obtained from the transport vehicles batteries can be used in ozone generators. The atmospheric air is possible to use as the ozone source.

II. EXPERIMENTAL

A schematic diagram of pilot plant is presented in Figure 1. In this experimental setup, the precise control of airflow rate, pollutant concentration, and reactor temperature and ozone concentration is possible. This setup can control these parameters accurately. Temperature conditions of the cleaning process were kept constant within a range of 25-400 °C. Toluene was used as a VOC (air contaminant). These were used as the catalysts (adsorbents): quartz, silica gel, aluminum oxide, spent platinum catalyst and natural zeolite (clinoptilolite).

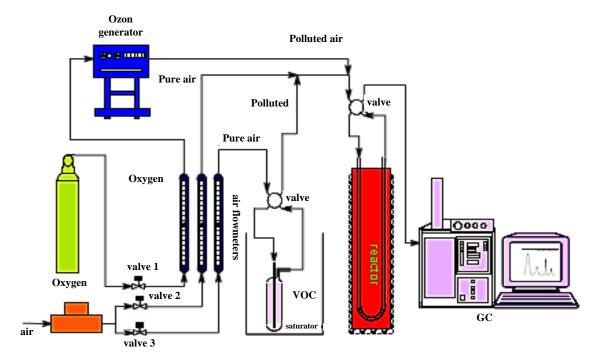


Figure 1. Schematic diagram of ozonation pilot plant

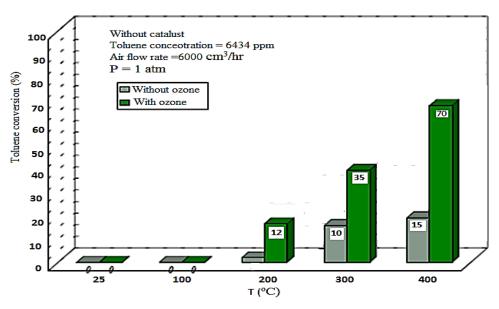


Figure 2. The polluted air cleaning without catalyst (adsorbent)

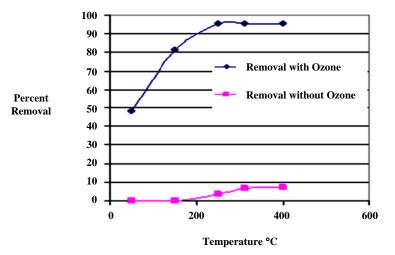


Figure 3. The polluted air cleaning over the quartz ($S = 3 \text{ m}^2$)

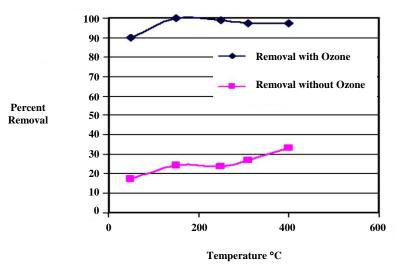


Figure 4. The polluted air cleaning over silica ($S = 120 \text{ m}^2$)

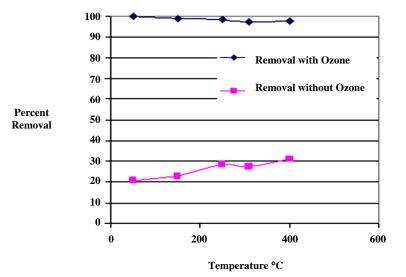
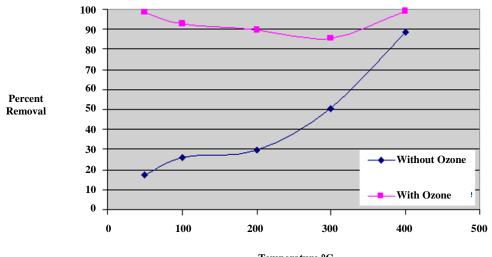


Figure 5. The polluted air cleaning over aluminum oxide ($S = 120 \text{ m}^2$)



Temperature °C

Figure 6. The polluted air cleaning over spent platinum catalyst ($S = 50 \text{ m}^2$)

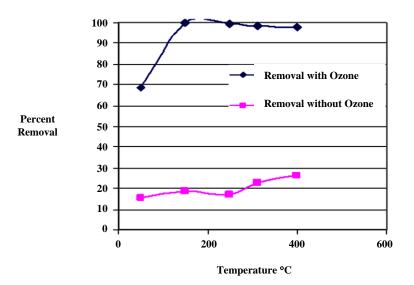


Figure 7. The polluted air cleaning over clinoptilolite ($S = 30 \text{ m}^2$)

III. RESULTS AND DISCUSSIONS

The obtained results are given in Figures 2 to 7. As can be seen from the given data, without employing a catalyst (adsorbent) toluene has not been removed completely (~70%). Treatment of the polluted air in the presence of a catalyst (adsorbent) at temperatures up to 400 °C also cannot remove all the VOCs completely. The maximum removal of VOCs over platinum catalyst was 88%. However, employing of catalysts (adsorbent) in the presence of ozone, the complete removal of toluene occurs at temperatures below 200 °C. The best result of toluene removal from the air is observed in the presence of ozone over platinum catalyst and aluminum oxide.

IV. CONCLUSIONS

1. The removal of carbon-containing compounds from the air using ozone as an oxidizer gives the best results in the presence of a catalyst (adsorbent).

2. As the atmospheric air can be used as a source of ozone, for cleaning of exhaust gases of transport vehicles the ozone generator working on the vehicles batteries should be used.

3. The optimal conditions of the cleaning process are to be determined for each catalyst (adsorbent).

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Anvar Teymur Khudiev was born in Fizuli, Azerbaijan on January 5, 1938. He received the M.Sc. degree from Azerbaijan State University (Baku, Azerbaijan) in 1961, the Ph.D. degree from Institute of Organic Chemistry Academy of Science (Moscow, USSR) in 1967 and Doctor of Science

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